

MATERIAL SAFETY DATA SHEET

:

Liqefed petroleum gas (LPG)

SECTION 1 - COMPANY INFORMATION

EMERGENCY TELEPHONE NUMBERS (24 hrs) COMPANY CONTACT (business hours) PSO House Control Room: (021)99203850 Manager HSE: (021)99207654, Ext: 2327 GM HSE: (021)99206939 Ext: 265 MSDS No. 7

ADDRESS:PSO House Khyaban-e-lqbal, Clifton, P.O.Box # 3983 Karachi 75600, Pakistan.Website:http://www.psopk.comSUPPLIERS:PRL, PARCO, OGDCL, PPL

	SECTION	2 – PRODU	JCT IDENTIFIC	CATION	I AND USE
Product name	Liquefied petroleum g	jas (LPG)	PIN (UN #)	1075
Chemical name	Liquefied petroleum g	TDG, DO	Г class	2.1	
Common names	Liquefied petroleum gas (LPG)		Packing	group	None
and synonyms	Compressed petroleum gas		Shipping	name	Liquefied petroleum gases;
Product use	Fuel, chemical interm			Or petroleum gases, liquefied	
WHMIS classification	Compressed Gas	Class A			
	Flammable Gas	Class B, Div	ision		
Hazard codes	NFPA Health	1 HMIS	Health	1	
	Flammability	4	Flammability	4	
	Reactivity	0	Reactivity	0	

NFPA & HMIS Ratings: 0=Insignificant/No Hazard. 1=Slight Hazard. 2=Moderate Hazard. 3=High/Serious Hazard. 4=Extreme/Severe Hazard.

SECTION 3 – HAZARDOUS INGREDIENTS							
Ingredients	CAS#	Wt %	ACGIH-TLVs (2004)	OSHA PELs (2004)	NIOSH RELs (2004)	LD ₅₀ (rat, oral)	LC₅₀ (rat, 4 hr)
LPG, which includes: Propane Butanes	106-97-8 74-98-6 106-97-8	100 95-100 <5	1000 ppm TWA 1000 ppm TWA 1000 ppm TWA	NAv 1000 ppm TWA NAv	800 ppm TWA 1000 ppm TWA 800 ppm TWA	NAp NAp NAp	658 mg/m ³ NAv 658 mg/m ³

SECTION 4 – PHYSICAL DATA				
Form	Gas. May be liquefied by pressurization	Vapour	853 kPa (8.4 at	m) at 21ºC
Colour	Colourless	Evaporation rate	NAp (gas)	
Odour	Very faint petroleum odour	Boiling point	-40° to 80 ° C (-40 to 176°F)	
Odour	22,000 to 36,000 mg/m ³	Freezing point	-190°C (-310°F) (propane)
Specific gravity	0.5 @ 20°C	РН	NAp	
Vapour density	1.55 (air = 1)	Coefficient of wa	ter/oil	2.36 [log P (oct)]

SECTION 5 - FIRE AND EXPLOSION HAZARDS Flammability X Yes 🗆 No Conditions Extremely flammable, explosive gas is released when liquid evaporates. Flash point -104°C (-156°F) (cc) Auto ignition 450°C (842°F) Lower flammable limit 1.8-1.9% (butane) Upper flammable limit 9.5% Static Accumulates static charge by flow or agitation. Ignites in Explosion data. Sensitivity Impac Not considered to: to be sensitive **discharge** response to static charge of sufficient energy. t

Means of extinctionStop flow. CO2 or dry chemical

Special precautions It is extremely dangerous to extinguish fire without stopping flow of gas. Gas and air will mix; resultant explosion could be more destructive than the original fire.

Gas is slightly heavier than air. It may travel a considerable distance to a source of ignition and flash back to a leak. Can accumulate in confined spaces, resulting in an explosion and/or asphyxiation

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Liqefed petroleum gas (LPG) MSDS Not hazard. Heating can cause rapid build-up of pressure inside containers, which may rupture

explosively. Hazardous combustion

Smoke. Carbon dioxide. Carbon monoxide.

products



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SECTION 6 – REACTIVITY INFORMATION					
Stability Stable Conditions to avoid Sources of ignition. Static discharges. High temperatures. Incompatible May react with strong oxidizing materials. Hazardous decomposition Carbon dioxide. Carbon monoxide.					
		SECTION 7 – HEALTH HAZ	ARD INFORM	ATION	
Route of Entry	 Eye Skin absorption Inhalation Ingestion 	Hazardous Cont on (w ith liquid fo	act Eye rm) Skin (Contact	
Acute exposure Contact with liquid will cause freeze injury to eyes or skin. High gas concentrations (% range) can cause asphyxiation by displacing air and thereby reducing the oxygen available for breathing. Symptoms include rapid breathing, fatigue, incoordination, excessive salivation, headache, nausea, vomiting and disorientation. Oxygen concentrations in work spaces must not be permitted to fall below 19%. Very high concentrations (10,000 ppm) can cause central nervous system (CNS) depression with symptoms such as headache, nausea, dizziness, drowsiness and confusion.					
Chronic	Not known to cause chronic effects.				
Carcinogenicity Not known to cause cancer. Not identified by OSHA as a carcinogen. Not classified by IARC, NTP, NIOSH, ACGIH, or EPA.					
Mutagenicity Teratogenicity Toxicologically sy products	Not known to be r NAv nergistic	nutagenic Simple asphyxiants (chemical:	Sensitization Reproductive s that displace	Not known to cause sensitization Not known to cause reproductive effects air in confined spaces).	
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SECTION 8 – FIRST AID

Inhalation Move affected person to fresh air or remove source of contamination. Give artificial respiration if breathing has stopped and if a qualified AR administrator is available. Obtain medical attention immediately.

- Ingestion Gases do not enter the body by this route.
- Eye Flush eye with lukewarm, gently flowing fresh water for at least 15 min. Do not attempt to re-warm. Cover both eyes with sterile dressing. Do not permit affected person to drink alcohol or smoke. Quickly transport affected person to an emergency medical facility.
- **Skin** Briefly flush the affected area with lukewarm, gently flowing water until the chemical is removed. Do not attempt to re-warm the affected area. Do not rub the affected area or apply dry heat. Carefully cut around clothing that sticks to the skin and remove the remainder of the garment. Loosely cover the affected area with a sterile dressing. Do not permit affected person to drink alcohol or smoke. Quickly transport affected person to an emergency medical facility.



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SECTION 9 – PRECAUTIONARY MEASURES						
Personal Protective Equipment	Gloves Eye Respiratory Clothing &	Insulated; preferably nitrile or Responder [™] Chemical safety glasses with face shield for working with the liquid product. NIOSH-approved SCBA or air line respirator with escape cylinder for work in confined spaces where oxygen deficiency may occur. A qualified occupational health and safety professional should be consulted for advice on respirator selection. Impervious protective clothing for total skin coverage when working with liquid product.				
Engineering controls Handling	Enclose product concentrations Eliminate all ign	Enclose product to the greatest extent possible. Use appropriate measures to ensure that oxygen concentrations do not fall below 19%. Eliminate all ignition sources. Use non-sparking equipment, explosion-proof ventilation systems, and				
procedures equipment Leak & spill procedure	 intrinsically safe electrical equipment. Bond and ground containers during product transfer. Have clean emergency eyewash and shower readily available in the work area. Evacuate area and keep it isolated until all gas has dispersed. Eliminate all sources of ignition. Ventila area. Stop leak if it can be done safely. Water spray may be used to dissipate gas. 					

Waste disposal	Consult local authorities. Controlled release to air may be permitted in certain situations.
Storage	Store in a cool, well-ventilated area. Keep away from strong oxidizing materials, excessive heat, and
	sources of ignition. Use non-sparking equipment, and explosion-proof ventilation systems. Consider leak
	detection and alarm equipment for storage area.
Shipping	Load at normal temperature (up to 38°C) and pressure. Bond and ground containers for transfer.

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